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# RIFM fragrance ingredient safety assessment, cinnamyl acetate, CAS Registry Number 103-54-8

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Name: Cinnamyl acetate  
CAS Registry Number: 103-54-8

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7. **Stereochemistry:** Isomer not specified. One geometric center present, and a total of two stereoisomers possible.

## 2. Physical data

1. **Boiling Point:** 113 °C at 3 mm Hg (Fragrance Materials Association [FMA]), 257.46 °C (EPI Suite), 269.0 °C (corrected to normal atmospheric pressure of 1013 hPa) (RIFM, 2017c)
2. **Flash Point:** >93 °C (Globally Harmonized System), >200 °F; CC (FMA), 130 °C (average corrected and rounded down to the nearest multiple of 0.5 °C) (RIFM, 2015c)
3. **Log K<sub>OW</sub>:** 2.7 at 35 °C (RIFM, 1999b), 2.85 (EPI Suite)
4. **Melting Point:** 20.45 °C (EPI Suite), -3.7 °C at 1005 hPa (RIFM, 2015b)
5. **Water Solubility:** 212.3 mg/L (EPI Suite)
6. **Specific Gravity:** 1.05 (FMA)
7. **Vapor Pressure:** 0.00751 mm Hg at 20 °C (EPI Suite v4.0), 0.008 mm Hg at 20 °C (FMA), 0.0121 mm Hg at 25 °C (EPI Suite)
8. **UV Spectra:** No significant absorbance between 290 and 700 nm; molar absorption coefficient is below the benchmark (1000 L mol<sup>-1</sup> • cm<sup>-1</sup>)
9. **Appearance/Organoleptic:** Colorless to slightly yellow, oily liquid with a sweet, balsamic, floral odor

## 3. Volume of use (worldwide band)

1. 10–100 metric tons per year (IFRA, 2015)

## 4. Exposure to fragrance ingredient (Creame RIFM aggregate exposure model v1.0)

1. **95th Percentile Concentration in Hydroalcohols:** 0.017% (RIFM, 2018b)
2. **Inhalation Exposure\*:** 0.00036 mg/kg/day or 0.027 mg/day (RIFM, 2018b)
3. **Total Systemic Exposure\*\*:** 0.0011 mg/kg/day (RIFM, 2018b)

\*95th percentile calculated exposure derived from concentration survey data in the Creame RIFM Aggregate Exposure Model (Comiskey, 2015, 2017; Safford, 2015, 2017).

\*\*95th percentile calculated exposure; assumes 100% absorption unless modified by dermal absorption data as reported in Section V. It is derived from concentration survey data in the Creame RIFM Aggregate Exposure Model and includes exposure via dermal, oral, and inhalation routes whenever the fragrance ingredient is used in products that include these routes of exposure (Comiskey, 2015, 2017; Safford, 2015, 2017).

## 5. Derivation of systemic absorption

1. **Dermal:** Assumed 100%
2. **Oral:** Assumed 100%
3. **Inhalation:** Assumed 100%

## 6. Computational toxicology evaluation

### 6.1. Cramer Classification

Class I, Low		
Expert Judgment	Toxtree v3.1	OECD QSAR Toolbox v4.2
I	I	I

### 6.2. Analogs Selected

- a. **Genotoxicity:** None
  - b. **Repeated Dose Toxicity:** None
  - c. **Reproductive Toxicity:** None
  - d. **Skin Sensitization:** None
  - e. **Phototoxicity/Photoallergenicity:** None
  - f. **Local Respiratory Toxicity:** None
  - g. **Environmental Toxicity:** None
3. **Read-across Justification:** None

## 7. Metabolism

Bickers (2005): Cinnamyl acetate is expected to be metabolized to cinnamyl alcohol and acetic acid (phase I metabolites).

Cinnamyl alcohol, cinnamaldehyde, and cinnamic acid are absorbed, metabolized, and excreted in the urine. They all follow the same metabolic pathway in that the alcohol is transformed into the aldehyde, which is metabolized to the acid. The final metabolite is hippuric acid, which is the principal metabolite being excreted in the urine (Fig. 1). The qualitative pattern of metabolism of cinnamaldehyde and cinnamic acid in humans is similar to that seen in laboratory species, and it is anticipated that this would also be broadly true for the metabolic fate of cinnamyl alcohol.

**Additional References:** None.

## 8. Natural occurrence

Cinnamyl acetate is reported by the VCF\* to occur in the following foods:

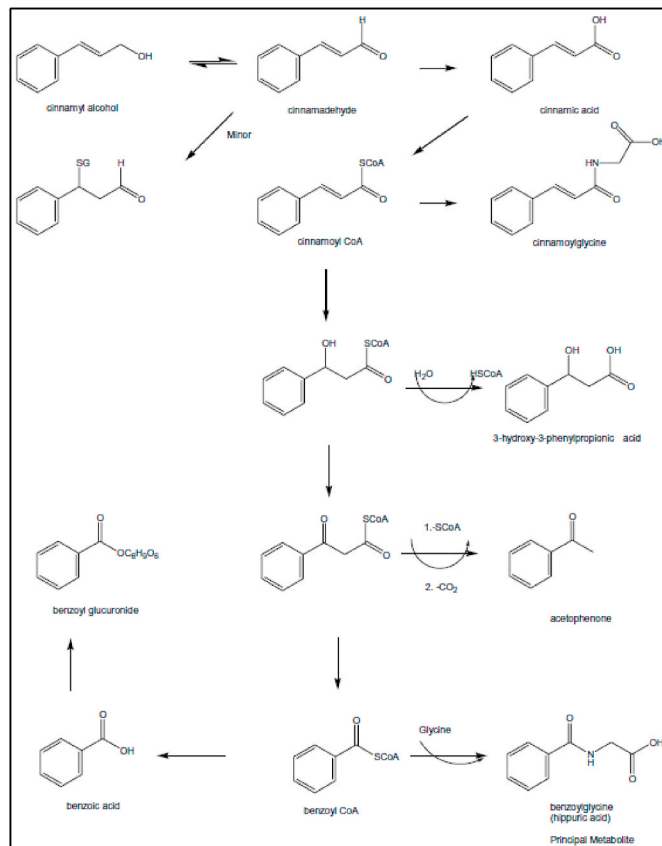


Fig. 1. Metabolism of cinnamyl alcohol, cinnamaldehyde, and cinnamic acid (adopted from Bickers, 2005).

Cinnamomum species  
 Guava and feyoa  
 Laurel (*Laurus nobilis* L.)  
 Litchi (*Litchi chinensis* Sonn.)  
 Melon  
 Ocimum species  
 Star anise  
 Starfruit (*Averrhoa carambola* L.)  
 Strawberry (*Fragaria* species)  
 Syzygium species  
 Tapereba, caja fruit (*Spondias lutea* L.)  
 Tarragon (*Artemisia dracunculus* L.)

\*VCF (Volatile Compounds in Food): Database/Nijssen, L.M.; Ingen-Visscher, C.A. van; Donders, J.J.H. (eds). – Version 15.1 – Zeist (The Netherlands): TNO Triskelion, 1963–2014. A continually updated database containing information on published volatile compounds that have been found in natural (processed) food products. Includes FEMA GRAS and EU-Flavis data.

## 9. REACH dossier

Available; accessed 04/08/21 (ECHA, 2016).

## 10. Conclusion

The existing information supports the use of this material as described in this safety assessment.

## 11. Summary

### 11.1. Human health endpoint summaries

#### 11.1.1. Genotoxicity

Based on the current existing data, cinnamyl acetate does not present a concern for genotoxicity.

**11.1.1.1. Risk assessment.** The mutagenic activity of cinnamyl acetate was assessed in an Ames study conducted in compliance with GLP requirements and in accordance with OECD TG 471 using both the standard plate incorporation and modified preincubation methods. *Salmonella typhimurium* strains TA1535, TA1537, TA98, TA100, and TA102 were treated with cinnamyl acetate in dimethyl sulfoxide (DMSO) at concentrations up to 5000 µg/plate in the presence and absence of metabolic activation. No increase in the number of revertant colonies was observed in any of the test strains at any concentration (RIFM, 2003). Under the conditions of the study, cinnamyl acetate was considered negative in the Ames test.

The clastogenic activity of cinnamyl acetate was evaluated in an *in vitro* micronucleus test conducted in compliance with GLP regulations and in accordance with OECD TG 487. Human peripheral blood lymphocytes were treated with cinnamyl acetate in DMSO at concentrations up to 1000 µg/mL in the presence and absence of metabolic activation at the 3-h and 24-h time points. A statistically significant increase in the frequency of binucleated cells with micronuclei (BNMN) was observed at the lowest evaluated concentration in the approximate 24-h treatment without S9. However, the BNMN frequency (1.10%) observed at this concentration was within the historical control range for this test condition and did not show a dose response; therefore, this increase was not considered biologically relevant. Cinnamyl acetate did not induce BNMN when tested up to the maximum dose in the 3-h treatments with and without metabolic activation (RIFM, 2015a). Under the conditions of the study, cinnamyl acetate was not considered clastogenic in the *in vitro* MNT assay. Chinese hamster ovary (CHO) cells were treated with cinnamyl acetate alone at doses of 1.0–100 µM; cells pretreated with

mitomycin C at 0.15 µM for 21 h were also treated with the test material. No increases in SCE frequency was observed in any of the test conditions at any concentration (Sasaki, 1989).

Based on the available data, cinnamyl acetate does not present a concern for genotoxic potential.

**Additional References:** None.

**Literature Search and Risk Assessment Completed On:** 08/21/20.

#### 11.1.2. Repeated dose toxicity

The MOE for cinnamyl acetate is adequate for the repeated dose toxicity endpoint at the current level of use.

**11.1.2.1. Risk assessment.** The repeated dose toxicity data on cinnamyl acetate are sufficient for the repeated dose toxicity endpoint. An OECD 422 and GLP-compliant, 28-day gavage combined repeated dose with a reproductive and developmental toxicity screening study was conducted with the test material. Groups of 10 Wistar rats/sex/dose were administered the test material via gavage at dose levels of 0, 65, 200, and 600 mg/kg/day in corn oil. An additional 14-day recovery group of 5 rats/sex assigned to the control and high-dose groups was also included. There were no treatment-related adverse effects reported among the treated animals up to the highest dose tested. Thus, the NOAEL for the repeated dose toxicity endpoint was determined to be 600 mg/kg/day, the highest dose tested (RIFM, 2016).

A default safety factor of 3 was used when deriving a NOAEL from the OECD 422 study (ECHA, 2012). The safety factor has been approved by the Expert Panel for Fragrance Safety\*.

Thus, the derived NOAEL for the repeated dose toxicity data is 600/3 or 200 mg/kg/day.

Therefore, the cinnamyl acetate MOE for the repeated dose toxicity endpoint can be calculated by dividing the cinnamyl acetate NOAEL in mg/kg/day by the total systemic exposure to cinnamyl acetate, 200/0.0011, or 181818.

In addition, the total systemic exposure to cinnamyl acetate (1.1 µg/kg/day) is below the TTC (30 µg/kg/day; Kroes, 2007) for the repeated dose toxicity endpoint of a Cramer Class I material at the current level of use.

\*The Expert Panel for Fragrance Safety is composed of scientific and technical experts in their respective fields. This group provides advice and guidance.

**Additional References:** None.

**Literature Search and Risk Assessment Completed On:** 08/13/20.

#### 11.1.3. Reproductive toxicity

The MOE for cinnamyl acetate is adequate for the reproductive toxicity endpoint at the current level of use.

**11.1.3.1. Risk assessment.** The reproductive toxicity data on cinnamyl acetate are sufficient for the developmental toxicity and fertility endpoints. An OECD 422 and GLP-compliant, 28-day gavage combined repeated dose with a reproductive and developmental toxicity screening study was conducted with the test material, cinnamyl acetate. Groups of 10 Wistar rats/sex/dose were administered the test material via gavage at dose levels of 0, 65, 200, and 600 mg/kg/day in corn oil. An additional 14-day recovery group of 5 rats/sex assigned to the control and high-dose groups were also included. The male and female mating and fertility indices were significantly lower at the 65 and 600 mg/kg/day doses when compared to controls. The changes observed at the 65 mg/kg/day dose were considered incidental as the observed changes were within the historical control data. The lower male and female mating and fertility indices at 600 mg/kg/day were considered treatment-related as the changes were lower than the historical control data. However, there were no effects of treatment on the reproductive organs



among the treated males and females. The pup survival index was also not altered by the treatment at all dose levels tested. No treatment-related developmental toxicity effects were reported among the treated animals up to the highest dose tested. Thus, the NOAEL for the developmental toxicity endpoint was determined to be 600 mg/kg/day, the highest dose tested. The NOAEL for the fertility endpoint was also determined to be 200 mg/kg/day, since male and female mating and fertility indices were significantly lower at the highest dose (RIFM, 2016).

Therefore, the cinnamyl acetate MOE for the developmental toxicity endpoint can be calculated by dividing the cinnamyl acetate NOAEL in mg/kg/day by the total systemic exposure to cinnamyl acetate, 600/0.0011, or 545455. The cinnamyl acetate MOE for the fertility endpoint can be calculated by dividing the cinnamyl acetate NOAEL in mg/kg/day by the total systemic exposure to cinnamyl acetate, 200/0.0011, or 181818.

In addition, the total systemic exposure to cinnamyl acetate (1.1 µg/kg/day) is below the TTC (30 µg/kg/day; Kroes, 2007; Laferrière, 2012) for the reproductive toxicity endpoint of a Cramer Class I material at the current level of use.

**Additional References:** None.

**Literature Search and Risk Assessment Completed On:** 08/12/20.

#### 11.1.4. Skin sensitization

Based on the existing data, cinnamyl acetate does not present a concern for skin sensitization.

**11.1.4.1. Risk assessment.** Based on existing data, cinnamyl acetate is not considered a skin sensitizer. The chemical structure of this material indicates that it would be expected to react with skin proteins (Roberts, 2007; Toxtree v3.1.0; OECD Toolbox v4.2). Cinnamyl acetate was found to be negative in the *in vitro* direct peptide reactivity assay (DPRA) (RIFM, 2017a) and human cell line activation test (h-CLAT) (RIFM, 2017b). In a human maximization test, no skin sensitization reactions were observed with 5% (3450 µg/cm<sup>2</sup>) cinnamyl acetate (RIFM, 1972). Additionally, in a Confirmation of No Induction in Humans test (CNIH) with 3424 µg/cm<sup>2</sup> of cinnamyl acetate in 1:3 ethanol:diethyl phthalate, no reactions indicative of sensitization were observed in any of the 101 volunteers (RIFM, 2018a).

Based on weight of evidence (WoE) from structural analysis and human studies, cinnamyl acetate does not present a concern for skin sensitization under the current, declared levels of use.

**Additional References:** None.

**Literature Search and Risk Assessment Completed On:** 08/05/20.

#### 11.1.5. Phototoxicity/photoallergenicity

Based on the available UV/Vis spectra, cinnamyl acetate would not be expected to present a concern for phototoxicity or photoallergenicity.

**11.1.5.1. Risk assessment.** There are no phototoxicity studies available for cinnamyl acetate in experimental models. UV/Vis absorption spectra indicate minor absorbance between 290 and 700 nm. The corresponding molar absorption coefficient is below the benchmark of concern for phototoxicity and photoallergenicity (Henry, 2009). Based on the lack of absorbance, cinnamyl acetate does not present a concern for phototoxicity or photoallergenicity.

**11.1.5.2. UV spectra analysis.** UV/Vis absorption spectra (OECD TG 101) were obtained. The spectra indicate minor absorbance in the range of 290–700 nm. The molar absorption coefficients (0, 0, and 43 L mol<sup>-1</sup> • cm<sup>-1</sup> under neutral, acidic, and basic conditions, respectively) are below the benchmark of concern for phototoxic effects, 1000 L mol<sup>-1</sup> • cm<sup>-1</sup> (Henry, 2009).

**Additional References:** None.

**Literature Search and Risk Assessment Completed On:** 08/10/20.

#### 11.1.6. Local Respiratory Toxicity

The MOE for cinnamyl acetate could not be calculated due to a lack of appropriate data. The exposure level for cinnamyl acetate is below the Cramer Class I TTC value for inhalation exposure local effects.

**11.1.6.1. Risk assessment.** There are insufficient inhalation data available on cinnamyl acetate. Based on the Creme RIFM Model, the inhalation exposure is 0.027 mg/day. This exposure is 51.9 times lower than the Cramer Class I TTC value of 1.4 mg/day (based on human lung weight of 650 g; Carthew, 2009); therefore, the exposure at the current level of use is deemed safe.

**Additional References:** Troy (1977); UGCM, 1997; Regnault-Roger (1995); Rice (1994); Kim (2004); Johnson (2005); Harth (2007); RIVM, 2007; RIFM, 2013; Carpenter (1949); DeCearrizz (1981); Brondeau (1990); Carlson (1946); Linyucheva (1971); Zissu (1995); Amdur (1961); Silver (1992); Zuskin (1997); Khare (1998); Helmig (1999a); Helmig (1999b); Montero (2001); Suzuki (2001); Morris (2002); Gagnaire (2002); NIOSH, 2006; Cain (2010); Willis (2011).

**Literature Search and Risk Assessment Completed On:** 08/19/20.

### 11.2. Environmental endpoint summary

#### 11.2.1. Screening-level assessment

A screening-level risk assessment of cinnamyl acetate was performed following the RIFM Environmental Framework (Salvito, 2002), which provides 3 tiered levels of screening for aquatic risk. In Tier 1, only the material's regional VoU, its log K<sub>ow</sub>, and its molecular weight are needed to estimate a conservative risk quotient (RQ), expressed as the ratio Predicted Environmental Concentration/Predicted No Effect Concentration (PEC/PNEC). A general QSAR with a high uncertainty factor applied is used to predict fish toxicity, as discussed in Salvito et al. (2002). In Tier 2, the RQ is refined by applying a lower uncertainty factor to the PNEC using the ECOSAR model (US EPA, 2012b), which provides chemical class-specific ecotoxicity estimates. Finally, if necessary, Tier 3 is conducted using measured biodegradation and ecotoxicity data to refine the RQ, thus allowing for lower PNEC uncertainty factors. The data for calculating the PEC and PNEC for this safety assessment are provided in the table below. For the PEC, the range from the most recent IFRA Volume of Use Survey is reviewed. The PEC is then calculated using the actual regional tonnage, not the extremes of the range. Following the RIFM Environmental Framework, cinnamyl acetate was identified as a fragrance material with the potential to present a possible risk to the aquatic environment (i.e., its screening-level PEC/PNEC >1).

A screening-level hazard assessment using EPI Suite v4.11 (US EPA, 2012a) did not identify cinnamyl acetate as possibly persistent or bioaccumulative based on its structure and physical-chemical properties. This screening-level hazard assessment considers the potential for a material to be persistent and bioaccumulative and toxic, or very persistent and very bioaccumulative as defined in the Criteria Document (Api, 2015). As noted in the Criteria Document, the screening criteria applied are the same as those used in the EU for REACH (ECHA, 2012). For persistence, if the EPI Suite model BIOWIN 3 predicts a value < 2.2 and either BIOWIN 2 or BIOWIN 6 predicts a value < 0.5, then the material is considered potentially persistent. A material would be considered potentially bioaccumulative if the EPI Suite model BCFBAF predicts a fish BCF ≥ 2000 L/kg. Ecotoxicity is determined in the above screening-level risk assessment. If, based on these model outputs (Step 1), additional assessment is required, a WoE-based review is then performed (Step 2). This review considers available data on the material's physical-chemical properties, environmental fate (e.g., OECD Guideline

biodegradation studies or die-away studies), fish bioaccumulation, and higher-tier model outputs (e.g., US EPA's BIOWIN and BCFBAF found in EPI Suite v4.11). Data on persistence and bioaccumulation are reported below and summarized in the Environmental Safety Assessment section prior to Section 1.

#### 11.2.2. Risk assessment

Based on the current Volume of Use (2015), cinnamyl acetate presents a risk to the aquatic compartment in the screening-level assessment.

##### 11.2.2.1. Key studies

**11.2.2.1.1. Biodegradation.** RIFM, 1999a: The ready biodegradability of the test material was determined by the manometric respirometry test according to OECD 301F guidelines. Biodegradation of 104% was observed after 28 days.

RIFM, 1999b: The degradation of the test material was evaluated using a closed bottle method according to OECD 301D guidelines. After 28 days, biodegradation of 75% was observed.

**11.2.2.1.2. Ecotoxicity.** RIFM, 1999c: A 48-h acute *Daphnia magna* study was conducted under static conditions. The EC50 was reported to be 29.4 mg/L.

**11.2.2.1.3. Other available data.** Cinnamyl acetate has been registered under REACH with the following additional data available at this time (ECHA, 2016):

The acute fish (*Danio rerio*) toxicity test was conducted according to the EU method C.1 under static conditions. The 96-h LC50 value based on mean measured concentrations was reported to be 2.76 mg/L.

The algae growth inhibition test was conducted according to the OECD 201 guidelines under static conditions. The 72-h EC50 value based on the time-weighted concentration was reported to be 12.3 mg/L (95% CI: 12–12.5 mg/L).

##### 11.2.3. Risk assessment refinement

Since cinnamyl acetate has passed the screening criteria, measured data is included in this document for completeness only and has not been used in PNEC derivation.

Ecotoxicological data and PNEC derivation (all endpoints reported in mg/L; PNECs in µg/L).

Endpoints used to calculate PNEC are underlined.

Exposure information and PEC calculation (following RIFM Environmental Framework: Salvito, 2002).

Exposure	Europe (EU)	North America (NA)
Log $K_{ow}$ Used	2.7	2.7
Biodegradation Factor Used	1	1
Dilution Factor	3	3
Regional Volume of Use Tonnage Band	10–100	1–10
<b>Risk Characterization: PEC/PNEC</b>	<b>&lt;1</b>	<b>&lt;1</b>

Based on available data, the RQ for this material is < 1. No further assessment is necessary.

The RIFM PNEC is 0.0932 µg/L. The revised PEC/PNECs for EU and NA are <1; therefore, the material does not present a risk to the aquatic environment at the current reported VoU.

**Literature Search and Risk Assessment Completed On: 08/20/20.**

## 12. Literature Search\*

- **RIFM Database:** Target, Fragrance Structure-Activity Group materials, other references, JECFA, CIR, SIDS
- **ECHA:** <https://echa.europa.eu/>
- **NTP:** <https://ntp.niehs.nih.gov/>
- **OECD Toolbox:** <https://www.oecd.org/chemicalsafety/risk-assessment/oecd-qsar-toolbox.htm>
- **SciFinder:** <https://scifinder.cas.org/scifinder/view/scifinder/scifinderExplore.jsf>
- **PubMed:** <https://www.ncbi.nlm.nih.gov/pubmed>
- **National Library of Medicine's Toxicology Information Services:** <https://toxnet.nlm.nih.gov/>
- **IARC:** <https://monographs.iarc.fr>
- **OECD SIDS:** <https://hvpchemicals.oecd.org/ui/Default.aspx>
- **EPA ACToR:** <https://actor.epa.gov/actor/home.xhtml>
- **US EPA HPVIS:** [https://ofmpub.epa.gov/opthpv/public\\_search\\_publicdetails?submission\\_id=24959241&ShowComments=Yes&sqlstr=null&recordcount=0&User\\_title=DetailQuery%20Results&EndPointRpt=Y#submission](https://ofmpub.epa.gov/opthpv/public_search_publicdetails?submission_id=24959241&ShowComments=Yes&sqlstr=null&recordcount=0&User_title=DetailQuery%20Results&EndPointRpt=Y#submission)
- **Japanese NITE:** [https://www.nite.go.jp/en/chem/chrip/chrip\\_search/systemTop](https://www.nite.go.jp/en/chem/chrip/chrip_search/systemTop)
- **Japan Existing Chemical Data Base (JECDB):** [http://dra4.nihs.go.jp/mhlw\\_data/jsp/SearchPageENG.jsp](http://dra4.nihs.go.jp/mhlw_data/jsp/SearchPageENG.jsp)

	LC50 (Fish) (mg/L)	EC50 ( <i>Daphnia</i> ) (mg/L)	EC50 (Algae) (mg/L)	AF	PNEC (µg/L)	Chemical Class
RIFM Framework Screening-level (Tier 1)	<u>58.48</u>			1000000	0.05848	
ECOSAR Acute Endpoints (Tier 2) v1.11	7.531	14.29	5.32			Esters

- **Google:** <https://www.google.com>
- **ChemIDplus:** <https://chem.nlm.nih.gov/chemidplus/>

Search keywords: CAS number and/or material names.

\*Information sources outside of RIFM's database are noted as appropriate in the safety assessment. This is not an exhaustive list. The links listed above were active as of 10/06/21.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper. We wish to confirm that there are no known conflicts of interest associated with this publication and there has been no significant financial support for this work that could have influenced its outcome. RIFM staff are employees of the Research Institute for Fragrance Materials, Inc. (RIFM). The Expert Panel receives a small honorarium for time spent reviewing the subject work.

## References

- Amdur, M.O., 1961. The respiratory response of Guinea pigs to the inhalation of acetic acid vapor. *Am. Ind. Hyg. Assoc. J.* 22 (1), 1–5.
- Api, A.M., Belsito, D., Bruze, M., Cadby, P., Calow, P., Dagli, M.L., Dekant, W., Ellis, G., Fryer, A.D., Fukayama, M., Griem, P., Hickey, C., Kromidas, L., Lalko, J.F., Liebler, D.C., Miyachi, Y., Politano, V.T., Renskers, K., Ritacco, G., Salvito, D., Schultz, T.W., Sipes, I.G., Smith, B., Vitale, D., Wilcox, D.K., 2015. Criteria for the Research Institute for fragrance materials, Inc. (RIFM) safety evaluation process for fragrance ingredients. *Food Chem. Toxicol.* 82, S1–S19.
- Bickers, D., Calow, P., Greim, H., Hanifin, J.M., Rogers, A.E., Saurat, J.H., Sipes, I.G., Smith, R.L., Tagami, H., 2005. A toxicologic and dermatologic assessment of cinnamyl alcohol, cinnamaldehyde and cinnamic acid when used as fragrance ingredients. *Food Chem. Toxicol.* 43 (6), 799–836.
- Brondeau, M.T., Bonnet, P., Guenier, J.P., Simon, P., DeCeuriz, J., 1990. Adrenal-dependent leucopenia after short-term exposure to various airborne irritants in rats. *J. Appl. Toxicol.* 10 (2), 83–86.
- Cain, W.S., Dourson, M.L., Kohrman-Vincent, M.J., Allen, B.C., 2010. Human chemosensory perception of methyl isothiocyanate: chemesthesis and odor. *Regul. Toxicol. Pharmacol.* 58 (2), 173–180.
- Carlson, G.W., 1946. Aplastic anemia following exposure to products of the sulfite pulp industry: a report of one case. *Ann. Intern. Med.* 24, 277–284.
- Carpenter, C.P., Smyth Jr., H.F., Pozzani, U.C., 1949. The assay of acute vapor toxicity, and the grading and interpretation of results on 96 chemical compounds. *J. Ind. Hyg. Toxicol.* 31 (6), 343–346.
- Carthew, P., Clapp, C., Gutsell, S., 2009. Exposure based waiving: the application of the toxicological threshold of concern (TTC) to inhalation exposure for aerosol ingredients in consumer products. *Food Chem. Toxicol.* 47 (6), 1287–1295.
- Comiskey, D., Api, A.M., Barratt, C., Daly, E.J., Ellis, G., McNamara, C., O'Mahony, C., Robison, S.H., Safford, B., Smith, B., Tozer, S., 2015. Novel database for exposure to fragrance ingredients in cosmetics and personal care products. *Regul. Toxicol. Pharmacol.* 72 (3), 660–672.
- Comiskey, D., Api, A.M., Barrett, C., Ellis, G., McNamara, C., O'Mahony, C., Robison, S.H., Rose, J., Safford, B., Smith, B., Tozer, S., 2017. Integrating habits and practices data for soaps, cosmetics and air care products into an existing aggregate exposure model. *Regul. Toxicol. Pharmacol.* 88, 144–156.
- De Ceauriz, J.C., Micillino, J.C., Bonnet, P., Guenier, J.P., 1981. Sensory irritation caused by various industrial airborne chemicals. *Toxicol. Lett.* 9, 137–143.
- ECHA, 2012. Guidance on Information Requirements and Chemical Safety Assessment. November 2012 v2.1. <http://echa.europa.eu/>.
- ECHA, 2016. Cinnamyl Acetate Registration Dossier. Retrieved from. <https://echa.europa.eu/lt/registration-dossier/-/registered-dossier/16797/1>.
- Gagnaire, F., Marignac, B., Hecht, G., Hery, M., 2002. Sensory irritation of acetic acid, hydrogen peroxide, peroxyacetic acid and their mixture in mice. *Ann. Occup. Hyg.* 46 (1), 97–102.
- Harth, V., Merget, R., Altmann, L., Bruning, T., 2007. Bronchial challenge testing to fragrance component as further diagnostic approach to non-immune immediate contact reactions. *Contact Dermatitis* 56 (3), 175–177.
- Helmig, D., Klinger, L.F., Guenther, A., Vierling, L., Geron, C., Zimmerman, P., 1999a. Biogenic volatile organic compound emissions (BVOCs). I. Identifications from three continental sites in the U.S. *Chemosphere* 38 (9), 2163–2187.
- Helmig, D., Klinger, L.F., Guenther, A., Vierling, L., Geron, C., Zimmerman, P., 1999b. Biogenic volatile organic compound emissions (BVOCs). II. Landscape flux potentials from three continental sites in the U.S. *Chemosphere* 38 (9), 2189–2204.
- Henry, B., Foti, C., Alsante, K., 2009. Can light absorption and photostability data be used to assess the photosafety risks in patients for a new drug molecule? *J. Photochem. Photobiol. B Biol.* 96 (1), 57–62.
- IFRA (International Fragrance Association), 2015. Volume of Use Survey. February 2015.
- Johnson, B.A., Farahbod, H., Leon, M., 2005. Interactions between odorant functional group and hydrocarbon structure influence activity in glomerular response modules in the rat olfactory bulb. *J. Comp. Neurol.* 483 (2), 205–216.
- Khare, P., Kumar, N., Satsangi, G.S., Maharaj Kumari, K., Srivastava, S.S., 1998. Formate and acetate in particulate matter and dust fall at Dayalbagh, Agra (India). *Chemosphere* 36 (14), 2993–3002.
- Kim, H.-K., Kim, J.-R., Ahn, Y.-J., 2004. Acaricidal activity of cinnamaldehyde and its congeners against tyrophagus putrescentiae (Acari: Acaridae). *J. Stored Prod. Res.* 40 (1), 55–63.
- Kroes, R., Renwick, A.G., Feron, V., Galli, C.L., Gibney, M., Greim, H., Guy, R.H., Lhuguenot, J.C., van de Sandt, J.J.M., 2007. Application of the threshold of toxicological concern (TTC) to the safety evaluation of cosmetic ingredients. *Food Chem. Toxicol.* 45 (12), 2533–2562.
- Laufersweiler, M.C., Gadagbui, B., Baskerville-Abraham, I.M., Maier, A., Willis, A., et al., 2012. Correlation of chemical structure with reproductive and developmental toxicity as it relates to the use of the threshold of toxicological concern. *Regul. Toxicol. Pharmacol.* 62 (1), 160–182.
- Linyucheva, L.A., 1971. Effects of the acetic acid inhalation on the citrate metabolism in the liver of albino rats. *Farmakologiya i Toksikologiya* 34 (4), 481–483.
- Montero, L., Vasconcellos, P.C., Souza, S.R., Pires, M.A.F., Andrade, O.R., Carvalho, L.R. F., 2001. Measurements of atmospheric carboxylic acids and carbonyl compounds in Sao Paulo City, Brazil. *Environ. Sci. Technol.* 35 (15), 3071–3081.
- Morris, J.B., Symanowicz, P.T., 2002. Immediate responses of mouse nose to inspired irritants. *Toxicologist* 66 (1-S), 98.
- Na, M., Ritacco, G., O'Brien, D., Lavelle, M., Api, A., Basketter, D., 2020. Fragrance skin sensitization evaluation and human testing, dermatitis. <https://doi.org/10.1097/DER.0000000000000684>. November 16, 2020. Volume Publish Ahead of Print Issue. Retrieved from.
- National Institute for Occupational Safety and Health, Kanwal, R., Kullman, G., Fedan, K., Kreiss, K., 2006. NIOSH Health Hazard Evaluation Report. Unpublished.
- Regnault-Roger, C., Hamraoui, A., 1995. Fumigant toxic activity and reproductive inhibition induced by monoterpenes on *Acanthoscelides obtectus* (Say) (Coleoptera), a bruchid of kidney bean (*Phaseolus vulgaris* L.). *J. Stored Prod. Res.* 31 (4), 291–299.
- Rice, P.J., Coats, J.R., 1994. Insecticidal properties of several monoterpenoids to the house fly (Diptera: muscidae), red flour beetle (Coleoptera: tenebrionidae) and southern corn rootworm (Coleoptera: chrysomelidae). *J. Econ. Entomol.* 87 (5), 1172–1179.
- RIFM (Research Institute for Fragrance Materials, Inc.), 1972. The Contact-Sensitization Potential of Fragrance Materials by Maximization Testing in Humans. Report to RIFM. RIFM Report Number 1804. RIFM, Woodcliff Lake, NJ, USA.
- RIFM (Research Institute for Fragrance Materials, Inc.), 1999a. Ready Biodegradability of Cinnamyl Acetate. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Givaudan. RIFM report number 51427.
- RIFM (Research Institute for Fragrance Materials, Inc.), 1999b. Partition Coefficient N-Octanol/water of Cinnamyl Acetate. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Givaudan. RIFM report number 51428.
- RIFM (Research Institute for Fragrance Materials, Inc.), 1999c. Cinnamyl Acetate: Acute Immobilisation Test with *Daphnia Magna* Straus and Biodegradation Study. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 57686.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2003. Evaluation of the Mutagenic Activity of Cinnamyl Acetate in the *Salmonella typhimurium* Reverse Mutation Assay. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Givaudan. RIFM report number 42155.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2013. Evaluation of Nose-Only Inhalation Exposure to Aerosolized Cinnamal in Sprague-Dawley Rats. RIFM, Woodcliff Lake, NJ, USA. RIFM report number 64503.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2015a. Cinnamyl Acetate: in Vitro Micronucleus Assay in Human Peripheral Blood Lymphocytes. RIFM Report Number 68481. RIFM, Woodcliff Lake, NJ, USA.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2015b. Cinnamyl Acetate: Determination of Physico-Chemical Properties Melting Point. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 70210.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2015c. Cinnamyl Acetate: Determination of Physico-Chemical Properties Flash Point. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 70212.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2016. Combined Repeated Dose Toxicity Study with the Reproduction/developmental Toxicity Screening Test with Cinnamyl Acetate (Cinnamylacetat) by Oral Gavage in Wistar Rats. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from RIFM report number 71062.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2017a. Cinnamyl Acetate (Cinnamylacetat): Direct Peptide Reactivity Assay. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 72219.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2017b. Cinnamyl Acetate (Cinnamylacetat): in Vitro Skin Sensitization Test - Human Cell Line Activation Test (H-CLAT). RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 72650.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2017c. Cinnamyl Acetate (Cinnamylacetat): Determination of Physico-Chemical Properties Boiling Point. RIFM, Woodcliff Lake, NJ, USA. Unpublished report from Symrise. RIFM report number 72801.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2018a. Cinnamyl Acetate: Repeated Insult Patch Test (RIPT). RIFM Report Number 73720. RIFM, Woodcliff Lake, NJ, USA.
- RIFM (Research Institute for Fragrance Materials, Inc.), 2018b. Expos. Survey 22. November 2018.
- RIVM., Rijksinstituut voor Volksgezondheid en M (National Institute for Public Health and Environment), Ezendam, J., De Klerk, A., Cassee, F.R., Van Loveren, H., de

- Jong, W.H., 2007. Immune Effects of Respiratory Exposure to Fragrance Chemicals. Pilot Studies with Isoeugenol and Cinnamal. Unpublished. RIVM Report 340301001/2007.
- Roberts, D.W., Patlewicz, G., Kern, P.S., Gerberick, F., Kimber, I., Dearman, R.J., Ryan, C. A., Basketter, D.A., Aptula, A.O., 2007. Mechanistic applicability domain classification of a local lymph node assay dataset for skin sensitization. *Chem. Res. Toxicol.* 20 (7), 1019–1030.
- Safford, B., Api, A.M., Barratt, C., Comiskey, D., Daly, E.J., Ellis, G., McNamara, C., O'Mahony, C., Robison, S., Smith, B., Thomas, R., Tozer, S., 2015. Use of an aggregate exposure model to estimate consumer exposure to fragrance ingredients in personal care and cosmetic products. *Regul. Toxicol. Pharmacol.* 72, 673–682.
- Safford, B., Api, A.M., Barratt, C., Comiskey, D., Ellis, G., McNamara, C., O'Mahony, C., Robison, S., Rose, J., Smith, B., Tozer, S., 2017. Application of the expanded Creme RIFM consumer exposure model to fragrance ingredients in cosmetic, personal care and air care products. *Regul. Toxicol. Pharmacol.* 86, 148–156.
- Salvito, D.T., Senna, R.J., Federle, T.W., 2002. A Framework for prioritizing fragrance materials for aquatic risk assessment. *Environ. Toxicol. Chem.* 21 (6), 1301–1308.
- Sasaki, Y.F., Imanishi, H., Phta, T., Shirasu, Y., 1989. Modifying effects of components of plant essence on the induction of sister-chromatid exchanges in cultured Chinese hamster ovary cells. *Mutat. Res. Lett.* 226 (1), 103–110.
- Silver, W.L., 1992. Neural and pharmacological basis for nasal irritation. In: *Annals of the New York Academy of Sciences*, vol. 641, pp. 152–163.
- Suzuki, Y., Kawakami, M., Akasaka, K., 2001. (1)H NMR Application for characterizing water-soluble organic compounds in urban atmospheric particles. *Environ. Sci. Technol.* 35 (13), 2656–2664.
- The Union of German Candle Manufacturers, 1997. Investigation of Oxidation Gases from Paraffin Aromatic Candles in Toxicological Relevance to Classes of Damaging Materials. Unpublished.
- Troy, W.R., 1977. Doctoral Dissertation: the Comparative Respiratory Irritation Potential of Fourteen Fragrance Raw Materials. Unpublished.
- US EPA, 2012a. Estimation Programs Interface Suite for Microsoft Windows, v4.0–v4.11. United States Environmental Protection Agency, Washington, DC, USA.
- US EPA, 2012b. The ECOSAR (ECOLOGical Structure Activity Relationship) Class Program for Microsoft Windows, v1.11. United States Environmental Protection Agency, Washington, DC, USA.
- Willis, D.N., Lin, B., Ha, M.A., Jordt, S.-E., Morris, J.B., 2011. Menthol attenuates respiratory irritation responses to multiple cigarette smoke irritants. *FASEB J.* 25 (12), 4434–4444.
- Zissu, D., 1995. Histopathological changes in the respiratory tract of mice exposed to ten families of airborne chemicals. *J. Appl. Toxicol.* 15 (3), 207–213.
- Zuskin, E., Mustajbegovic, J., Schachter, E.N., Pavicic, D., Budak, A., 1997. A follow-up study of respiratory function in workers exposed to acid aerosols in a food-processing industry. *Int. Arch. Occup. Environ. Health* 70, 413–418.