

Short review

RIFM fragrance ingredient safety assessment, α -Methylbenzyl acetate, CAS Registry Number 93-92-5

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ABSTRACT

The use of this material under current use conditions is supported by the existing information.

This material was evaluated for genotoxicity, repeated dose toxicity, developmental toxicity, reproductive toxicity, local respiratory toxicity, phototoxicity, skin sensitization potential, as well as, environmental safety. Developmental toxicity was determined to have the most conservative systemic exposure derived NO[A]EL of 100 mg/kg/day. A gavage developmental toxicity study conducted in rats on a suitable read across analog resulted in aMOE of 3571 while considering 78.7% absorption from skin contact and 100% from inhalation. A MOE of >100 is deemed acceptable.

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1. Identification

1. **Chemical Name:** α -Methylbenzyl acetate

2. **CAS Registry Number:** 93-92-5

3. **Synonyms:** α -Methylbenzyl acetate, Benzenemethanol, α -methyl-, acetate, Gardenol, Methylphenylcarbinyl acetate, sec-Phenylethyl acetate, α -Phenylethyl acetate, Phenyl methyl

carbinyl acetate, Styralyl acetate, アルキル(C = 1–5)カルボン酸71
ニルアルキル(C = 1–6), 1-Phenylethyl acetate

4. **Molecular Formula:** C₁₀H₁₂O₂

5. **Molecular Weight:** 164.2

6. **RIFM Number:** 178

2. Physical data

1. **Boiling Point:** 214 °C [FMA database], 223.12 °C [EPI Suite]

2. **Flash Point:** 195 °F; CC [FMA database]

3. **Log Kow:** 2.5 at 30 °C [RIFM, 1996b], 2.5 [EPI Suite]

4. **Melting Point:** –0.17 °C [EPI Suite]

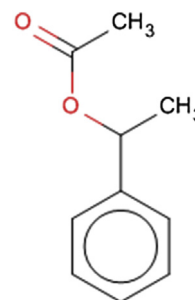
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Version: 082415. This version replaces any previous versions.

Name: α -Methylbenzyl acetate

CAS Registry Number: 93-92-5



Abbreviation list:

2-Box Model – a RIFM, Inc. proprietary *in silico* tool used to calculate fragrance air exposure concentration

97.5th percentile- The concentration of the fragrance ingredient is obtained from examination of several thousand commercial fine fragrance formulations. The upper 97.5th percentile concentration is calculated from these data and is then used to estimate the dermal systemic exposure in ten types of the most frequently used personal care and cosmetic products. The dermal route is the major route in assessing the safety of fragrance ingredients. Further explanation of how the data were obtained and of how exposures were determined has been previously reported by Cadby et al. (2002) and Ford et al. (2000).

AF- Assessment Factor

BCF- Bioconcentration Factor

DEREK- Derek nexus is an *in silico* tool used to identify structural alerts

DST- Dermal Sensitization Threshold

ECHA-European Chemicals Agency

EU – Europe/European Union

GLP- Good Laboratory Practice

IFRA- The International Fragrance Association

LOEL- Lowest Observable Effect Level

MOE- Margin of Exposure

MPPD - Multiple-Path Particle Dosimetry. An *in silico* model for inhaled vapors used to simulate fragrance lung deposition

NA – North America

NESIL- No Expected Sensitization Induction Level

NOAEC- No Observed Adverse Effect Concentration

NOAEL- No Observed Adverse Effect Level

NOEC- No Observed Effect Concentration

OECD- Organisation for Economic Co-operation and Development

OECD TG- Organisation for Economic Co-operation and Development Testing Guidelines

PBT- Persistent, Bioaccumulative, and Toxic

PEC/PNEC- Predicted Environmental Concentration/Predicted No Effect Concentration

QRA- quantitative risk assessment

REACH- Registration, Evaluation, Authorisation, and Restriction of Chemicals

RIFM- Research Institute for Fragrance Materials

RQ- Risk Quotient

TTC- Threshold of Toxicological Concern

UV/Vis Spectra- Ultra Violet/Visible spectra

VCF- Volatile Compounds in Food

VoU- Volume of Use

vPvB- (very) Persistent, (very) Bioaccumulative

WOE – Weight of Evidences

RIFM's Expert Panel* concludes that this material is safe under the limits described in this safety assessment.

This safety assessment is based on the RIFM Criteria Document (Api et al., 2015) which should be referred to for clarifications.

Each endpoint discussed in this safety assessment reviews the relevant data that were available at the time of writing (version number in the top box is indicative of the date of approval based on a two digit month/day/year), both in the RIFM database (consisting of publicly available and proprietary data) and through publicly available information sources (i.e., SciFinder and PubMed). Studies selected for this safety assessment were based on appropriate test criteria, such as acceptable guidelines, sample size, study duration, route of exposure, relevant animal species, most relevant testing endpoints, etc. A key study for each endpoint was selected based on the most conservative end-point value (e.g., PNEC, NOAEL, LOEL, and NESIL).

5 **Water Solubility:** 481.1 mg/L [EPI Suite]

6 **Specific Gravity:** 1.023–1.026 [FMA database], 1.025–1.028 [FMA], 1.0241 [RIFM Database], 1.03 g/ml [RIFM, 1994], 1.023–1.026 @ 25/25 °C [Gaunt et al., 1974]

7 **Vapor Pressure:** 5.5 Pa at 20 °C [RIFM, 2011], 0.0733 mm Hg @ 20 °C [EPI Suite 4.0], 0.1 mm Hg 20 °C [FMA database], 0.112 mm Hg @ 25 °C [EPI Suite]

8 **UV Spectra:** No significant absorption in the region of 290–700 nm; molar absorption coefficient is below the benchmark ($1000 \text{ L mol}^{-1} \text{ cm}^{-1}$).

9 **Appearance/Organoleptic:** A clear, colorless to pale yellow liquid, having an intense green odor suggesting gardenia.

*RIFM's Expert Panel is an independent body that selects its own members and establishes its own operating procedures. The Expert Panel is comprised of internationally known scientists that provide RIFM guidance relevant to human health and environmental protection.

Summary: The use of this material under current use conditions is supported by the existing information.

This material was evaluated for genotoxicity, repeated dose toxicity, developmental toxicity, reproductive toxicity, local respiratory toxicity, phototoxicity, skin sensitization potential, as well as, environmental safety. Developmental toxicity was determined to have the most conservative systemic exposure derived NO[A]EL of 100 mg/kg/day. A gavage developmental toxicity study conducted in rats on a suitable read across analog resulted in a MOE of 3571 while considering 78.7% absorption from skin contact and 100% from inhalation. A MOE of >100 is deemed acceptable.

Human Health Safety Assessment

Genotoxicity: Not genotoxic	(Tennant et al., 1987; Shelby et al., 1993)
Repeated Dose Toxicity: NOAEL = 150 mg/kg/day	(Gaunt et al., 1974)
Developmental and Reproductive Toxicity: NOAEL = 100 mg/kg/day	(Ishiguro et al., 1993)
Skin Sensitization: Not sensitizing	(RIFM, 1985c; RIFM, 1986a; RIFM, 1985d; RIFM, 1987; RIFM, 1988a; RIFM, 1961; RIFM, 1975a; RIFM, 1975b; RIFM, 1975c; RIFM, 1975d; RIFM, 1975e; RIFM, 1988b; RIFM, 1988c; RIFM, 1988d; Klecak, 1985; RIFM, 1970; NTP, 1993; RIFM, 1985b)
Phototoxicity/Photoallergenicity: Not phototoxic/ photoallergenic	(UV Spectra, RIFM Database)
Local Respiratory Toxicity: NOAEC = 10 ppm or 61.4 mg/ m ³	(RIFM, 2013b)

Environmental Safety Assessment

Hazard Assessment:

Persistence: Critical Measured Value: 106% (OECD 301B)	(RIFM, 1996a)
Bioaccumulation: Screening Level: 20.68 L/kg	(EpiSuite ver 4.1)
Ecotoxicity: Critical Ecotoxicity Endpoint: 96 h fish (<i>Brachydanio rerio</i>) LC50: 21.0 mg/l	(RIFM, 1993)
Conclusion: Not PBT as per IFRA Environmental Standards	
Risk Assessment:	
Screening-Level: PEC/PNEC (North America and Europe) > 1	(RIFM Framework; Salvito et al., 2002; http://rifmdatabase.rifm.org/RifmDatabase/Studies/40315)
Critical Ecotoxicity Endpoint: 96 h fish (<i>Brachydanio rerio</i>) LC50: 21.0 mg/l	(RIFM, 1993)
RIFM PNEC is: 3.66 µg/L	
•Revised PEC/PNECs (2011 IFRA VoU): North America and Europe <1	

3. Exposure

estimated from the amount of ¹⁴C-equivalents excreted in the urine over the 5 day collection period. When the application site was occluded with either plastic wrap or a glass chamber, the

1. Volume of Use (worldwide band): >1000 metric tons per year	(IFRA, 2011)
2. Average Maximum Concentration in Hydroalcohols: 1.2%	(IFRA, 2008)
3. 97.5th Percentile: 1.28%	(IFRA, 2008)
4. Dermal Exposure*: 0.0326 mg/kg/day	(IFRA, 2008)
5. Oral Exposure: Not available	
6. Inhalation Exposures**: 0.002 mg/kg/day	(IFRA, 2008)
7. Total Systemic Exposure (Dermal + Inhalation): (0.0326 mg/kg/day × 78.7% absorption) + 0.002 mg/kg/day = 0.028 mg/kg/day	

*Calculated using the reported 97.5th percentile concentration based on the levels of the same fragrance ingredient in ten of the most frequently used personal care and cosmetic products (i.e., anti-perspirant, bath products, body lotion, eau de toilette, face cream, fragrance cream, hair spray, shampoo, shower gel, and toilet soap) (Cadby et al., 2002; Ford et al., 2000).

**Combined (fine fragrances, hair sprays, antiperspirants/deodorants, candles, aerosol air fresheners, and reed diffusers/heated oil plug-ins) result calculated using RIFM's 2-Box/MPPD *in silico* models, based on the IFRA survey results for the 97.5th percentile use in hydroalcohols for a 60 kg individual.

4. Derivation of systemic absorption

1 **Dermal:** 78.7%, read-across from benzyl acetate (CAS # 140-11-4) (Bronaugh et al., 1990: The skin absorption of read across material [7-¹⁴C] benzyl acetate (CAS # 104-11-1; see Section 5) was measured in 4 female rhesus monkeys. The test material in acetone was applied at a concentration of 4 µg/cm² to a 1 cm² area of abdominal skin for 24 h. Urine was collected for an additional 4 days. The extent of dermal absorption was

absorption of benzyl acetate was 17.3 ± 2.7% and 78.7 ± 7.5%, respectively. When the site was not occluded, the absorption was 34.6 ± 9.4%.

2 **Oral:** Data not available – not considered.

3 **Inhalation:** Assumed 100%

4 **Total:** Dermal (78.7%) + Inhalation (assume 100%) absorbed = (0.0326 mg/kg/day × 78.7%) + 0.002 mg/kg/day = 0.028 mg/kg/day

5. Computational toxicology evaluation

1 Cramer Classification: Class I, Low

Expert judgment	Toxtree v 2.6	OECD QSAR Toolbox v 3.2
I	I	I

2 Analogues Selected:

- a Genotoxicity: Benzyl acetate (CAS # 140-11-4)
 - b Repeated Dose Toxicity: benzyl acetate (CAS # 140-11-4)
 - c Developmental and Reproductive Toxicity: Benzyl acetate (CAS # 140-11-4)
 - d Skin Sensitization: Benzyl acetate (CAS # 140-11-4)
 - e Phototoxicity/Photoallergenicity: None
 - f Local Respiratory Toxicity: Benzyl acetate (CAS # 140-11-4)
 - g Environmental Toxicity: None
- 3 Read-across Justification: See [Appendix](#) below

6. Metabolism

Not considered for this risk assessment and therefore not reviewed except where it may pertain in specific endpoint sections as discussed below.

7. Natural occurrence (discrete chemical) or composition (Ncs)

α -Methylbenzyl acetate is reported to occur in the following foods*:

Avocado (*Persea americana* Mill.)

Guava and Feyoa.

Strawberry guava (*Psidium cattleianum* Sabine).

*VCF Volatile Compounds in Food: database/Nijssen, L.M.; Ingen-Visscher, C.A. van; Donders, J.J.H. [eds]. – Version 15.1 – Zeist (The Netherlands): TNO Triskelion, 1963–2014. A continually updated database, contains information on published volatile compounds which have been found in natural (processed) food products. Includes FEMA GRAS and EU-Flavis data.

8. IFRA standard

None.

9. REACH dossier

Available; accessed on 04/23/14.

10. Summary

10.1. Human health endpoint summaries

10.1.1. Genotoxicity

Based on the current existing data and use levels, α -methylbenzyl acetate does not present a concern for genetic toxicity.

10.1.1.1. Risk assessment. α -Methylbenzyl acetate was tested by the BlueScreen assay and was found negative for both cytotoxicity and genotoxicity indicating a lack of genotoxic potential (RIFM, 2013a).

There are no studies assessing the mutagenic potential of α -methylbenzyl acetate, however, read across can be made to benzyl acetate (CAS # 140-11-4; See Section 5) which was assessed for mutagenicity in an Ames study similar to OECD TG 471 using the plate incorporation method. *S. typhimurium* strains TA98, TA100, TA1535, and TA1537 were exposed to concentrations of benzyl acetate up to 10 mg per plate in the presence and absence of liver S-9 fractions. No significant increases in revertant colonies were seen with benzyl acetate with or without metabolic activation (S9). The study concluded that benzyl acetate is not mutagenic under the conditions of this test (Tennant et al., 1987).

There are no studies assessing the clastogenic activity of α -methylbenzyl acetate. The clastogenic potential of read across analog benzyl acetate was assessed in several *in vitro* mouse lymphoma cell assays with and without metabolic activation with varying results. However, several *in vivo* studies assessing the effect of the material on inducing chromosomal aberrations, unscheduled DNA synthesis, demonstrate that benzyl acetate lacks genotoxic potential (NTP, 1993; Steinmetz and Mirsalis, 1984). In an *in vivo* mammalian erythrocyte micronucleus assay performed in equivalence to or similar to OECD TG 474, groups of 5–7 male mice were administered benzyl acetate in corn oil via intraperitoneal injection for 3 consecutive days at doses up to 1250 mg/kg. No genotoxic activity was observed, and the material was considered not clastogenic (Shelby et al., 1993).

Taken together, benzyl acetate does not present a concern for genotoxic potential and this can be extended to α -methylbenzyl acetate.

Additional References: NTP, 1993; Florin et al., 1980; Mortelmans et al., 1986; Schunk et al., 1986; Rogan et al., 1986; Mirsalis et al., 1989; Steinmetz and Mirsalis, 1984; Mirsalis et al., 1983; Foureman et al., 1994; Matsuoka et al., 1996; Yoshikawa, 1996; Miyagawa et al., 1995; Mitchell and Caspary, 1987; Zimmermann, 1989; Honma et al., 1999; Kevekordes et al., 1999; Rossman et al., 1991; Kevekordes et al., 2001; Sekihashi et al., 2002; Yoo, 1985; Demir et al., 2010; Scott et al., 2007; Yasunaga et al., 2004; Witt et al., 2000; Sasaki et al., 2000; Oda et al., 1978; Elmore and Fitzgerald, 1990; Longnecker et al., 1990; Galloway et al., 1987; Caspary et al., 1988; Rudd et al., 1983; Yoo, 1986; McGregor et al., 1988.

Literature Search and Risk Assessment Completed on: 05/07/14.

10.1.2. Repeated dose toxicity

The margin of exposure for α -methylbenzyl acetate is adequate for the repeated dose toxicity endpoint at the current level of use.

10.1.2.1. Risk assessment. The repeated dose toxicity data on α -methylbenzyl acetate are sufficient for the repeated dose toxicity endpoint. A gavage 13-week subchronic toxicity study conducted in rats determined the NOAEL to be 150 mg/kg/day, the highest dosage tested (Gaunt et al., 1974). **Therefore, the MOE is equal to the NOAEL in mg/kg/day divided by the total systemic exposure, 150/0.028 or 5357.**

Additional References: McGinty et al., 2012a; Belsito et al., 2012; McGinty et al., 2012b, 2012c, 2012d; McGinty, 2012e; RIFM, 2013b; RIFM, 1986b; RIFM, 1957; Abdo and Wenk, 1995; Abdo et al., 1998; Longnecker et al., 1986, 1990; Young, 1989; Abdo et al., 1985; Caldwell et al., 1987; Snapper et al., 1925; Hotchkiss et al., 1992a; Nasser-Sina et al., 1992; Chidgey et al., 1986;

Grundschober, 1977; Miyashita and Robinson, 1980; Chidgey and Caldwell, 1986; Chidgey et al., 1987; McMahon et al., 1989a; Augustinsson and Ekedahl, 1962; Clapp and Young, 1970; McMahon et al., 1989b; Schunk et al., 1986; RIFM, 1989b; Hotchkiss et al., 1992b; Caldwell et al., 1987; Hotchkiss, 1998; Hotchkiss et al., 1992c; Meyer, 1965; Garnett et al., 1994; Jimbo, 1983; Hotchkiss et al., 1988, 1990a, 1990b, 1989; RIFM, 1989a; Hotchkiss et al., 1992d.

Literature Search and Risk Assessment Completed on: 05/02/14.

10.1.3. Developmental and reproductive toxicity

The margin of exposure for α -methylbenzyl acetate is adequate for the developmental and reproductive toxicity endpoints at the current level of use.

10.1.3.1. Risk assessment. There are no developmental toxicity data on α -methylbenzyl acetate. Read across material benzyl acetate (CAS # 140-11-4; see Section 5) has a gavage developmental toxicity study that was conducted in rats. The NOAEL for developmental toxicity was determined to be 100 mg/kg/day, based on minor fetal internal anomalies and fetal weights (Ishiguro et al., 1993). **Therefore, the MOE for developmental toxicity is equal to the benzyl acetate NOAEL in mg/kg/day divided by the total systemic exposure, 100/0.028 or 3571.**

There are no reproductive toxicity data on α -methylbenzyl acetate. Read across material benzyl acetate (CAS # 140-11-4) has a gavage developmental toxicity study conducted in rats that determined the NOAEL for maternal toxicity to be 500 mg/kg/day, based on maternal body weight gain (Ishiguro et al., 1993). In 13-week dietary subchronic toxicity studies in rats and mice with benzyl acetate, sperm morphology and vaginal cytology examinations were evaluated (Morrissey et al., 1988; NTP, 1993). There were no effects on sperm parameters in mice or rats up to the high dosage of 7900 or 3900 mg/kg/day, respectively. There were no effects on estrous cycling in female rats up to the high dosage of 4500 mg/kg/day. Lengthening of the estrous cycle occurred in high-dose female mice (9400 mg/kg/day), which the authors concluded was related to decreases in body weight. These data indicate no specific concern for reproductive toxicity. **Therefore, the MOE for reproductive toxicity is equal to the benzyl acetate NOAEL in mg/kg/day divided by the total systemic exposure, 500/0.028 or 17857.**

RIFM's Expert Panel* and the adjunct Reproduction Advisory Group reviewed the Ishiguro et al. (1993) results and concluded that the minor fetal anomalies observed at the highest dose level were most likely a developmental delay. An increased fetal body weight may indicate an adverse effect. For example, chemicals which induce maternal diabetes mellitus may increase fetal weight. Early embryocidal effects leading to a reduced litter size may secondarily increase fetal weight. The Panel members and the Reproduction Advisory Group concluded that the increased fetal body weight observed in the 100 and 10 mg/kg/day groups was biologically insignificant and that no additional reproductive or developmental toxicity studies are needed. They concluded that the maternal NOAEL was 500 mg/kg/day based on weight gain and the fetal NOAEL is 100 mg/kg/day based on weight and internal organ malformations.

*RIFM's Expert Panel and adjunct Reproduction Advisory Group are composed of scientific and technical experts in their respective

fields. These groups provide advice and guidance.

Additional References: McGinty et al., 2012a; Belsito et al., 2012; McGinty et al., 2012b, 2012c, 2012d; McGinty, 2012e; RIFM, 2013b; RIFM, 1986b; RIFM, 1957; Abdo and Wenk, 1995; Abdo et al., 1998; Longnecker et al., 1986, 1990; Young, 1989; Abdo et al., 1985; Caldwell et al., 1987; Snapper et al., 1925; Hotchkiss et al., 1992a; Nasser-Sina et al., 1992; Chidgey et al., 1986; Grundschober, 1977; Miyashita and Robinson, 1980; Chidgey and Caldwell, 1986; Chidgey et al., 1987; McMahon et al., 1989a; Augustinsson and Ekedahl, 1962; Clapp and Young, 1970; McMahon et al., 1989b; Schunk et al., 1986; RIFM, 1989b; Hotchkiss et al., 1992b; Caldwell et al., 1987; Hotchkiss, 1998; Hotchkiss et al., 1992c; Meyer, 1965; Garnett et al., 1994; Jimbo, 1983; Hotchkiss et al., 1988, 1990a, 1990b, 1989; RIFM, 1989a; Hotchkiss et al., 1992d.

Literature Search and Risk Assessment Completed on: 05/02/14.

10.1.4. Skin sensitization

Based on existing material specific data and read across to benzyl acetate (CAS # 140-11-4), α -methylbenzyl acetate does not present a concern for skin sensitization.

10.1.4.1. Risk assessment. Based on existing material specific data and read across to benzyl acetate (CAS # 140-11-4; see Section 5), α -methylbenzyl acetate does not present a concern for skin sensitization. While the chemical structure of these materials indicates that they would be expected to react with skin proteins, the reactivity would be expected to be low (Roberts et al., 2007; Toxtree 2.5.0; OECD toolbox v3.1). In guinea pig test methods no reactions indicative of sensitization were observed to benzyl acetate (RIFM, 1985d; RIFM, 1986a; RIFM, 1985c; RIFM, 1985a; Klecak, 1985 RIFM, 1985b). Additionally, no reactions indicative of skin sensitization were observed in both the human repeated insult patch test and the human maximization test to benzyl acetate (RIFM, 1988a; RIFM, 1961; RIFM, 1975a; RIFM, 1975b; RIFM, 1975c; RIFM, 1975d; RIFM, 1975e; RIFM, 1988b; RIFM, 1988c; RIFM, 1988d; NTP, 1993, RIFM, 1987). Finally, in the human maximization test no reactions indicative of sensitization were observed to α -methylbenzyl acetate (RIFM, 1970).

Additional References: None.

Literature Search and Risk Assessment Completed on: 05/05/14.

10.1.5. Phototoxicity/photoallergenicity

Based on UV/Vis absorption spectra, α -methylbenzyl acetate would not be expected to present a concern for phototoxicity or photoallergenicity.

10.1.5.1. Risk assessment. There are no phototoxicity studies available for α -methylbenzyl acetate in experimental models. UV/Vis absorption spectra indicate no significant absorption between 290 and 700 nm. Corresponding molar absorption coefficient is well below the benchmark of concern for phototoxicity and photoallergenicity 1000 L mol⁻¹ cm⁻¹ (Henry et al., 2009). Based on lack of absorbance, α -methylbenzyl acetate does not present a concern for phototoxicity or photoallergenicity.

Additional References: None.

Literature Search and Risk Assessment Completed on: 08/21/15.

10.1.6. Local respiratory toxicity

The margin of exposure for α -methylbenzyl acetate is adequate for the respiratory endpoint at the current level of use.

10.1.6.1. Risk assessment. The inhalation exposure estimated for combined exposure was considered along with toxicological data observed in the scientific literature to calculate the MOE from inhalation exposure when used in perfumery. There are no inhalation data available on α -methylbenzyl acetate. A NOAEC of 10 ppm (61.4 mg/m³) is reported for read across analog, benzyl acetate (CAS # 140-11-4; see Section 5), for a 2 week acute study conducted in rats (RIFM, 2013b). At this level, increased lactate dehydrogenase was noted in the bronchoalveolar lavage fluid. Although the authors did not consider these effects as adverse, for the purpose of estimating local respiratory toxicity MOE, the lower exposure dose (61.4 mg/m³) was considered.

This NOAEC expressed in mg/kg lung weight/day is:

- (61.4 mg/m³) (1m³/1000 L) = 0.0614 mg/L
- Minute ventilation (MV) of 0.17 L/min for a Sprague–Dawley rat \times duration of exposure of 360 min per day (min/day) (according to GLP study guidelines) = 61.2 L/d
- (0.0614 mg/L) (61.2 L/d) = 3.76 mg/d
- (3.76 mg/d)/(0.0016 kg lung weight of rat*) = 2349 mg/kg lw/day

Based on the IFRA survey results for hydroalcoholics the 97.5th percentile was reported to be 1.28%. Assuming the same amount is used in all product types (fine fragrances, hair sprays, anti-perspirants/deodorants, candles, aerosol air fresheners, and reed diffusers/heated oil plug-ins) the combined inhalation exposure would be 0.12 mg/day as calculated using RIFM's 2-Box/MPPD *in silico* models, based on the IFRA survey results for the 97.5th percentile use in hydroalcoholics for a 60 kg individual. To compare this estimated exposure with the NOAEC expressed in mg/kg lung weight/day this value is divided by 0.65 kg human lung weight (Carthew et al., 2009) to give 0.181 mg/kg lung weight/day resulting in a MOE of 129,765 (i.e., [23487.5 mg/kg lw/day]/[0.181 mg/kg lung weight/day]).

The MOE is significantly greater than 100. Without the adjustment for specific uncertainty factors related to inter-species and intra-species variation the material exposure by inhalation at 1.28% in a combination of the products noted above is deemed to be safe under the most conservative consumer exposure scenario.

Additional References: RIFM, 1977; RIFM, 1997b; Silver, 1992; RIFM, 1997a; Isola et al., 2003a; RIFM, 2003b; Rogers et al., 2003; RIFM, 2003a; Isola et al., 2003b; Isola et al., 2004a; Smith et al., 2004; RIFM, 2004; Isola et al., 2004b; Rogers et al., 2005; Randazzo et al., 2014; Vethanayagam et al., 2013.

Literature Search and Risk Assessment Completed on: 09/23/14.

10.2. Environmental endpoint summary

10.2.1. Screening-level assessment

A screening level risk assessment of α -methylbenzyl acetate was performed following the RIFM Environmental Framework (Salvito et al., 2002) which provides for 3 levels of screening for aquatic risk. In Tier 1, only the material's volume of use in a region, its log K_{ow} and molecular weight are needed to estimate a

conservative risk quotient (RQ; Predicted Environmental Concentration/Predicted No Effect Concentration or PEC/PNEC). In Tier 1, a general QSAR for fish toxicity is used with a high uncertainty factor as discussed in Salvito et al., 2002. At Tier 2, the model ECOSAR (providing chemical class specific ecotoxicity estimates) is used and a lower uncertainty factor is applied. Finally, if needed, at Tier 3, measured biodegradation and ecotoxicity data are used to refine the RQ (again, with lower uncertainty factors applied to calculate the PNEC). Following the RIFM Environmental Framework, α -methylbenzyl acetate was identified as a fragrance material with the potential to present a possible risk to the aquatic environment (i.e., its screening level PEC/PNEC >1).

A screening-level hazard assessment using EPISUITE ver 4.1 did not identify α -methylbenzyl acetate as either being possibly persistent nor bioaccumulative based on its structure and physical–chemical properties. This screening level hazard assessment is a weight of evidence review of a material's physical–chemical properties, available data on environmental fate (e.g., OECD Guideline biodegradation studies or die-away studies) and fish bioaccumulation, and review of model outputs (e.g., USEPA's BIOWIN and BCFBAF found in EPISUITE ver.4.1). Specific key data on biodegradation and fate and bioaccumulation are reported below and summarized in the Environmental Safety Assessment section prior to Section 1.

10.2.2. Risk assessment

Based on current volume of use (2011), α -methylbenzyl acetate presents a risk to the aquatic compartment in the screening level assessment.

10.2.3. Key studies

10.2.3.1. Biodegradation. RIFM, 1994: The ready and ultimate biodegradability of α -methylbenzyl acetate was determined by a CO₂ production test based on OECD Guideline 301B. α -Methylbenzyl acetate at 10 mg/l organic carbon was directly added to the incubation mixture, and incubated for 28 days. Biodegradation was 106.0% (102.3%–109.6%).

RIFM, 1996: The Ready Biodegradability of the test material was determined by the Manometric Respirometry Test following the OECD 301F method. Under the conditions of this study, biodegradation of 89% was observed.

RIFM, 1993: The biodegradation potential of the test material was measured using the CO₂ evolution method following the OECD 301B guidelines. After 28 days of incubation, a biodegradation of 65% was observed.

10.2.3.2. Ecotoxicity. RIFM, 1993: A 96 h acute fish (*Brachydanio rerio*) toxicity test was conducted with the test material. Under conditions of the study the LCO, LC50 and LC100 were determined to be 15.0, 21.0 and 28.5 mg/l, respectively.

10.2.3.3. Other available data. α -Methylbenzyl acetate has been registered under REACH, however no additional data is available.

11. Risk assessment refinement

In the REACH dossier PNEC has been calculated using the same key study (fish acute) as in this document, but since read – across data for additional endpoints is also included the Assessment Factor is lower (1000 vs 5000).

Ecotoxicological data and PNEC derivation (all endpoints

	LC50 (Fish)	EC50 (Daphnia)	EC50 (Algae)	AF	PNEC	Chemical Class
RIFM Framework Screening Level (Tier 1)	<u>81.35</u> mg/l			1,000,000	0.08135 µg/l	
ECOSAR Acute Endpoints (Tier 2) Ver 1.11	11.23 mg/l	22.13 mg/l	<u>8.703 mg/l</u>	10,000	0.8703 µg/l	Esters
ECOSAR Acute Endpoints (Tier 2) Ver 1.11	48.11 mg/l	28.60 mg/l	25.77 mg/l			Neutral Organics
Tier 3: Measured Data including REACH						
	LC50	EC50	NOEC	AF	PNEC	Comments
Fish	21.0 mg/l			5000	3.66 µg/l	
Daphnia						
Algae						

reported in mg/L; PNECs in µg/L).

Endpoints used to calculate PNEC are underlined.

Exposure	Europe (EU)	North America (NA)
Log K _{ow} used	2.5	2.5
Biodegradation Factor Used	1	1
Dilution Factor	3	3
Regional Volume of Use Tonnage Band	100–1000	100–1000
Risk Characterization: PEC/PNEC	<1	<1

Exposure information and PEC calculation (following RIFM Framework: [Salvito et al., 2002](#)).

Based on available data, the RQ for this material is < 1. No additional assessment is necessary.

The RIFM PNEC is 3.66 µg/L. The revised PEC/PNECs for EU and NA are <1 and therefore, do not present a risk to the aquatic environment at the current reported volumes of use.

Literature Search and Risk Assessment Completed on: 5/06/14.

12. Literature Search*

- **RIFM database:** target, Fragrance Structure Activity Group materials, other references, JECFA, CIR, SIDS
- **ECHA:** <http://echa.europa.eu/>

- **NTP:** http://tools.niehs.nih.gov/ntp_tox/index.cfm
- **OECD Toolbox**
- **SciFinder:** <https://scifinder.cas.org/scifinder/view/scifinder/scifinderExplore.jsf>
- **PUBMED:** <http://www.ncbi.nlm.nih.gov/pubmed>
- **TOXNET:** <http://toxnet.nlm.nih.gov/>
- **IARC** (<http://monographs.iarc.fr/>):
- **OECD SIDS:** <http://www.chem.unep.ch/irptc/sids/oecdsids/sidspub.html>
- **EPA Actor:** <http://actor.epa.gov/actor/faces/ACToRHome.jsp?jsessionid=0EF5C212B7906229F477472A9A4D05B7>
- **US EPA HPVIS:** <http://www.epa.gov/hpv/hpvis/index.html>
- **US EPA Robust Summary:** <http://cfpub.epa.gov/hpv-s/>
- **Japanese NITE:** <http://www.safe.nite.go.jp/english/db.html>
- **Japan Existing Chemical Data Base:** http://dra4.nihs.go.jp/mhlw_data/jsp/SearchPageENG.jsp
- **Google:** <https://www.google.com/webhp?tab=ww&ei=KMSoUpiQK-arsQS324GwBg&ved=0CBQQ1S4>

*Information sources outside of RIFM's database are noted as appropriate in the safety assessment.

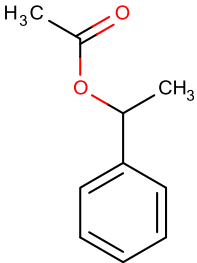
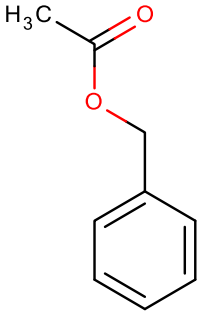
This is not an exhaustive list.

Appendix A. Supplementary data

Transparency document

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.fct.2016.01.020>.

Transparency document related to this article can be found online at <http://dx.doi.org/10.1016/j.fct.2016.01.020>.

	Target material	Read across material
Principal Name	alpha-Methylbenzyl acetate	Benzyl acetate
CAS No.	93-92-5	140-11-4
Structure		
3D Structure	http://www.thegoodscentscompany.com/opl/93-92-5.html	http://www.thegoodscentscompany.com/opl/140-11-4.html
Molecular Formula:	C ₁₀ H ₁₂ O ₂	C ₉ H ₁₀ O ₂
Molecular Weight:	164.21	150.17
Melting Point (°C)	16.86	-0.5
Boiling Point (°C)	234.30	215.57
Vapor Pressure (mmHg @ 25°C)	0.0624	0.187
Log Kow (KOWWIN v1.68 estimate)	2.63	2.08
Water Solubility (mg/L, @ 25°C, WSKOW v1.42 estimate)	539.8	1605
J_{max} (mg/cm²/h, RIFM SAM)	74.6	64.0
Henry's Law (Pa·m³/mol, Bond Method)	1.58	1.4337
Similarity (Tanimoto score)¹		58%
Skin Absorption		
Skin Absorption Percentage (SAM)	80%	80%
Genotoxicity		
DNA binding by OASIS v 1.1		
QSAR Tool Box (3.1)	<ul style="list-style-type: none"> Schiff base formers Schiff base formers >> Direct acting Schiff base formers Schiff base formers >> Direct acting Schiff base formers >> Specific Acetate Esters SN1 SN1 >> Carbenium ion formation SN1 >> Carbenium ion formation >> Specific Acetate Esters SN2 SN2 >> Acylating agents SN2 >> Acylating agents >> Specific Acetate Esters SN2 >> SN2 at sp3-carbon atom SN2 >> SN2 at sp3-carbon atom >> Specific Acetate Esters Michael addition Michael addition >> P450 Mediated Activation to Quinones and Quinone-type Chemicals Michael addition >> P450 Mediated Activation to Quinones and Quinone-type Chemicals >> Arenes No alerts 	<ul style="list-style-type: none"> Schiff base formers Schiff base formers >> Direct acting Schiff base formers Schiff base formers >> Direct acting Schiff base formers >> Specific Acetate Esters SN1 SN1 >> Carbenium ion formation SN1 >> Carbenium ion formation >> Specific Acetate Esters SN2 SN2 >> Acylating agents SN2 >> Acylating agents >> Specific Acetate Esters SN2 >> SN2 at sp3-carbon atom SN2 >> SN2 at sp3-carbon atom >> Specific Acetate Esters Michael addition Michael addition >> P450 Mediated Activation to Quinones and Quinone-type Chemicals Michael addition >> P450 Mediated Activation to Quinones and Quinone-type Chemicals >> Arenes No alerts
DNA binding OECD		
Carcinogenicity (genotox and non-genotox) alerts by ISS		
DNA alerts for Ames, MN, CA by OASIS v 1.1	<ul style="list-style-type: none"> Schiff base formers Schiff base formers >> Direct acting Schiff base formers Schiff base formers >> Direct acting Schiff base formers >> Specific Acetate Esters SN1 SN1 >> Carbenium ion formation SN1 >> Carbenium ion formation >> Specific Acetate Esters SN2 SN2 >> Acylating agents SN2 >> Acylating agents >> Specific Acetate Esters SN2 >> SN2 at sp3-carbon atom SN2 >> SN2 at sp3-carbon atom >> Specific Acetate Esters No alerts 	<ul style="list-style-type: none"> Schiff base formers Schiff base formers >> Direct acting Schiff base formers Schiff base formers >> Direct acting Schiff base formers >> Specific Acetate Esters SN1 SN1 >> Carbenium ion formation SN1 >> Carbenium ion formation >> Specific Acetate Esters SN2 SN2 >> Acylating agents SN2 >> Acylating agents >> Specific Acetate Esters SN2 >> SN2 at sp3-carbon atom SN2 >> SN2 at sp3-carbon atom >> Specific Acetate Esters No alerts
In-vitro Mutagenicity (Ames test) alerts by ISS	<ul style="list-style-type: none"> H-acceptor-path3-H-acceptor 	<ul style="list-style-type: none"> H-acceptor-path3-H-acceptor

(continued)

Target material	Read across material
In-vivo mutagenicity (Micronucleus) alerts by ISS	
Oncologic Classification	• Not classified
Reproductive and Developmental toxicity	
Developmental Toxicity Model by CAESAR v2.1.6	• Non toxicant (low reliability)
ER Binding by OECD QSAR Toolbox (3.1)	• Non binder
Skin sensitization	
Protein binding by OASIS v1.1	• SN2 • SN2 >> Nucleophilic substitution at sp3 Carbon atom • SN2 >> Nucleophilic substitution at sp3 Carbon atom >> Activated alkyl esters
Protein binding by OECD	• SN2 • SN2 >> SN2 reaction at sp3 carbon atom • SN2 >> SN2 reaction at sp3 carbon atom >> Allyl acetates and related chemicals
Protein binding potency	• Not possible to classify according to these rules (GSH)
Protein binding alerts for skin sensitization by OASIS v1.1	• SN2 • SN2 >> Nucleophilic substitution at sp3 Carbon atom • SN2 >> Nucleophilic substitution at sp3 Carbon atom >> Activated alkyl esters
Skin Sensitization model (CAESAR) (version 2.1.5)	• Sensitizer (low reliability)
Metabolism	
OECD QSAR Toolbox (3.1)	Supplemental data 1
Rat liver S9 metabolism simulator	Supplemental data 2

¹ Values calculated using Pipeline Pilot with FCFP fingerprint (Rogers and Hahn, 2010).

Appendix B.

Summary:

There are insufficient toxicity data on alpha-methylbenzyl acetate (RIFM# 178, CAS# 93-92-5). Hence, *in silico* evaluation was conducted to determine suitable read-across material. Based on structural similarity, reactivity, metabolism data, physicochemical properties and expert judgment, the above shown read-across materials were identified as proper read across for their respective toxicity endpoints.

Methods:

- The identified read-across analogs were confirmed by using expert judgment
- The physicochemical properties of target and analog were calculated using EPI Suite™ v4.11 developed by US EPA (USEPA, 2012)
- The J_{max} were calculated using RIFM skin absorption model (SAM), the parameters were calculated using consensus model (Shen et al., 2014)
- DNA binding, mutagenicity, genotoxicity alerts and oncologic classification were estimated using OECD QSAR Toolbox (v3.1) (OECD, 2012)
- Repeat dose categorization were estimated using OECD QSAR Toolbox (v3.1) (OECD, 2012)
- Skin sensitization were estimated using CAESAR (v2.1.6) (Cassano et al., 2010)
- Protein binding were estimated using OECD QSAR Toolbox (v3.1) (OECD, 2012)
- The major metabolites for the target and read-across analogs were determined and evaluated using OECD QSAR Toolbox (v3.1) (OECD, 2012)

Conclusion/Rationale

- Benzyl acetate (analog) was used as a read-across for alpha-methylbenzyl acetate (target) based on:
 - o The target and analog belong to the generic class of aromatic esters, specifically, esters/aryl alkyl alcohol simple acid/benzylic alcohol.
 - o The target and analog have the same carboxylic acid part and similar alcohol part.
 - o The only difference is that the target has a methyl group attached to the alpha carbon in the alcohol part. The difference between structures does not essentially change the physicochemical properties nor raise any additional structural alerts and therefore, their toxicology profiles are expected to be similar.
 - o The target and analog show similar alerts for DNA binding, mutagenicity, genotoxicity and oncologic classification.
 - o The target and analog show similar alerts for Repeated Dose (HESS) Categorization and ER Binding. ER Binding is molecular initiating event analogous to protein binding. ER binding is not necessarily predictive of endocrine disruption given the complex pre- and post-receptor events that determine activity.
 - o The target and analog show similar alerts for protein binding.
 - o The target and analog are expected to be metabolized similarly. As per the OECD Toolbox they are predicted to have similar metabolites.

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